# Mylonitization And Decomposition Of Garnet: Evidence For Rapid Deformation And Entrainment Of Mantle Garnet-Harzburgite By Kimberlite Magma, K1 Pipe, Venetia Mine, South Africa

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#### Abstract

Sheared and unsheared nodules of garnet-harzburgite from the K1 Venetia kimberlite pipe (South Africa) are composed primarily of olivine and orthopyroxene (partially replaced by serpentine, magnetite and chlorite) and abundant porphyroblasts of garnet with kelyphitic rims of orthopyroxene, clinopyroxene and spinel. When mylonitized, orthopyroxene defines a mineral elongation lineation while bent orthopyroxene and sigmoidal garnet with wings of orthopyroxene define consistent senses of motion within individual nodules. The kelyphite surrounding both undeformed and undeformed garnet grains is not deformed and appears to have resulted from the inversion of garnet and olivine to orthopyroxene, clinopyroxene and spinel, corresponding to the isograd separating garnet peridotite from spinel peridotite. Subsequent hydration around nodule rims and along fractures within nodules resulted in the formation of serpentine, magnetite and chlorite from olivine and orthopyroxene. This hydration is believed to have resulted from reaction with the entraining kimberlite magma. The preservation of mylonite rather than its annealing and recrystallisation to coarse-grained rocks at high temperature strain free mantle conditions requires quenching shortly after mylonite formation. This preservation taken with the presence of undeformed kelyphite around garnet grains implies that mylonitization and entrainment of garnet-harzburgite into kimberlite magma and kelyphite formation occurred during very rapid magma ascent.

# Introduction

At the present level of exposure (10 level, ~110 m depth), garnet-harzburgite is the most common mantle nodule composition found within the ~530 Ma (Allsopp *et al.*, 1995) K1 kimberlite pipe at the Venetia Mine, South Africa (Figures 1 and 2). Associated primarily

with the earliest phase of hypabyssal kimberlite intrusion as deduced from cross cutting relationships (*e.g.* Seggie *et al.*, 1999) (Figure 3), these nodules range in texture from massive and unsheared to mylonitic (Figure 4). Abundant garnet, whether deformed or not, is always rimmed by undeformed kelyphite (Figure 4E), showing that kelyphite formation postdated deformation. In addition, reaction rims between kimberlite magma and the nodules surround only some of the kelyphitic rims, showing the kelyphite formation is not a result of chemical interaction with kimberlite magma. In this paper we document the nature of mylonitization of garnet-harzburgite, the subsequent generation of kelyphitic rims around undeformed and deformed garnet and then hydration from the kimberlite magma. It is argued that these processes are associated with the rapid, forceful passing of kimberlitic magma through actively deforming garnet-harzburgite on the way to the surface, and then decompression of entrained nodules of this composition.

#### The K1 Kimberlite Pipe

The K1 kimberlite pipe (Figures 2 and 3) is one of 13 bodies, exposed at surface or blind, presently recognized within the Venetia kimberlite cluster (*e.g.* Seggie *et al.*, 1999). It intruded into >2.0 Ga gneisses and schists of the Central Zone of the Limpopo Belt (*e.g.* Pienaar, 1985; S. Pretorius, 1986; 1992; W. Pretorius, 1996) as well as gabbroic sills and lavas and bodies of sodic pegmatite of various ages between 1.6 and 1.8 Ga age (Pretorius, 1996; Barton and Pretorius, 1997; Twiggs *et al.*, 2002). Composed of "monticellite-phlogopite kimberlite of Group I character", it comprises six hypabyssal and diatreme or tuffaceous kimberlite breccia (TKB) phases (Seggie *et al.*, 1999). While mantle and crustal nodules occur within all six of these phases, they are by far most abundant in earliest hypabyssal-facies phase (H-N and H-S; Figure 3). The distribution of deep mantle nodules within the kimberlite phases of the K1 pipe is heterogeneous. The collection at the Rand Afrikaans University (RAU), which was sampled primarily from seven level downwards, is dominated by garnet-harzburgite (>80% of ~ 300 nodules) with a few examples of harzburgite, pyroxenite, dunite and eclogite. The smaller collection studied by Stieffenhoffer *et al.* (1998; 1999) at the DeBeers Geoscience Centre (Johannesburg, South Africa) was collected from seven level upwards and is dominated by garnet-lherzolite with lesser amounts of garnet-harzburgite, garnet-spinel peridotite and spinel peridotite, the latter two types presumably from a lower P-T environment.

#### **Garnet-Harzburgite Nodules**

After examining thin sections of the entire RAU nodule collection, six nodules of garnet-harzburgite were selected for detailed petrographic study (Figure 4) on the basis of their distinct microstructural features and a low degree of hydration (Table 1). Undeformed to moderately deformed garnet-harzburgite nodules are large ellipsoids or fragments thereof, ranging in size up to 40 x 30 x 20 cm (B-00-184, Figure 4D). As the garnet-harzburgite becomes progressively sheared, its nodules become smaller and more flattened, *e.g.* sample B-00-136, 10 x 5 x 3 cm (Figure 4F). The long and short axes of the sheared nodules define the planes containing the mineral elongation lineations.

The nodules of undeformed garnet-harzburgite are coarse-grained, porphyroblastic rocks composed primarily of large (5 to 10 mm) fractured grains of olivine and 1 to 5 mm isometric grains of pyroxenes with 5 to 10 mm euhedral garnet porphyroblasts. The pyrox-enes are mostly orthopyroxene although relatively rare grains of euhedral clinopyroxene occur

up to 0.3 cm in long axis. Most clinopyroxene is associated with kelyphitic rims around garnet (see below). Fracturing of olivine grains is accompanied by the development of serpentine in the cracks. Fracturing is not accompanied by the significant displacement of olivine segments. Garnet porphyroblasts are always surrounded by radial kelyphitic aggregates (Figure 5a) composed of pyroxenes and spinel. Similar symplectites also fill cracks in the garnet. The degree of replacement of garnet varies from 10 to 100% and does not depend on the location of the garnet porphyroblasts relatively to the xenolith margins.

In comparison to undeformed nodules of the same mineral composition, deformed garnet-harzburgite nodules are commonly characterized by smaller grain size and a distinct orientation of elongated orthopyroxene grains defining the shear fabric (Figure 6). Reduction of grain size is very characteristic for intensely deformed mylonitized rocks, which also contain abundant shear sense indicators such as bending of the pyroxene grains (Figures 6c and d) and rarely by sigmoid garnet (Figure 6a) and orthopyroxene (Figure 6b) porphyroblasts. Olivine grains in sheared xenoliths are intensely fractured and fragmented (Figures 6b and c). In less deformed rocks olivine fragments are elongated and orientated along the fabric. As with the undeformed nodules, garnet grains are surrounded by 0.1 to 1 mm, radial kelyphytic rims (Figure 5b) that can replace up to 100% of the grains independent of location within the nodule. The radial internal texture of kelyphite clearly suggests their post-shear origin. Cracks in olivine are filled by serpentine. In intensely mylonitized samples, serpentine often composes a significant part of the rock matrix surrounding abundant small random fragments of olivine (Figure 6c).

Darker rims, up to 5 cm wide, occur around the edges of the nodules whether deformed or not and reflect increased serpentine and chlorite and decreased olivine and orthopyroxene content. They resulted from reaction with fluids presumably from the kimberlite magma during transport. These rims surround coarsely crystalline garnet with kelyphitic rims and orthopyroxene crystals without apparent reaction.

#### **Reaction textures**

Two types of reaction textures are distinguished in all samples studied: 1) Replacement of garnet by orthopyroxene-clinopyroxene-spinel kelyphite (Figure 5) and 2) Subsequent hydration textures (replacement of olivine by serpentine and magnetite and orthopyroxene by chlorite) superimposed on both initial mineral assemblages (Figures 6b and c) and kelyphitic textures (Figure 7).

The internal structures of kelyphitic rims around garnet are characterized by relatively coarse-grained (5 to 30  $\mu$ m) outer portions with orthopyroxene, clinopyroxene and spinel without preferred mineral orientation, surrounding cryptocrystalline (<3  $\mu$ m) radial aggregates of the same minerals replacing garnet (Figure 5). Formation of kelyphitic textures appears to be controlled by diffusion of the components on the garnet-olivine boundaries and related to the retrograde reaction garnet plus olivine goes to orthopyroxene, clinopyroxene and spinel, corresponding to the isograd between garnet peridotite and spinel peridotite assemblages. This reaction seems to be near isochemical in a local scale and not related to significant addition of the components from kimberlitic magma. However, in one instance, a Ba rich phase also occurs within the orthopyroxene-clinopyroxene-spinel kelyphite (Figure 5d). Presently, the origin and significance of the Ba is not understood but Ba-bearing phase may be a distinct hint on metasomatism caused by kimberlitic magma.

Hydration reaction textures are very common for all types of nodules. Although in the inner portions of nodules these reactions are mostly developed along the grain boundaries and cracks inside the minerals, the complete replacement of olivine to serpentine and magnetite and orthopyroxene to chlorite occurs in the altered outer (black) zones of the nodules. These reactions are likely to have resulted from the infiltration of a water bearing fluid during the transport of nodules within kimberlitic magma.

#### **Composition of minerals**

Microprobe analyses of coexisting minerals from studied samples were carried out using the CAMECA microprobe and the Scanning Electronic Microscope in the Faculty of Science at Rand Afrikaans University as well as the Electronic Microprobe of the Institute of Geology, Mineralogy and Geophysics at Ruhr-University of Bochum. Table 2 contains representative microprobe analyses of minerals from both deformed (B-00-136, B-00-184) and undeformed (VN17, VN19) samples. Major characteristics of chemical compositions of garnets and pyroxenes from studied samples are summarized in Figures 8 and 9. Most of the matrix minerals are characterized by homogeneous chemical compositions (Figures 8a and b; 9a and b) and by the absence of chemical zoning (Figure 8b). However in the moderately deformed sample B-00-184, systematic zoning of Cr contents is observed in garnet (Figure 8d) and pyroxenes (Table 2).

Compositions of minerals within kelyphitic textures are inhomogeneous and strong zoning of Al and Cr contents is common in relatively large (10 to  $30 \,\mu$ m) grains of spinel and pyroxenes composing the outer portions of the symplectites (Figure 9d, Table 2). However, the distribution of Ca, Mg and Fe between clinopyroxene and orthopyroxene is very uniform

for all grains studied and is less variable then those detected for matrix minerals (compare Figures 9a and c).

#### Thermobarometry of the nodules

#### Methodology

Thermobarometry of mantle nodule is recognized as a complex problem (see review by Smith, 1999) that cannot be solved by simple routine methods applicable for example to lower grade metamorphic rocks (e.g. Frost and Chacko, 1989; Spear and Florence, 1992; Spear, 1993). Several authors (e.g., Harley, 1984; Frost and Chacko, 1989; Spear and Florence, 1992; Spear, 1993) have also noted that geothermometers based on Fe-Mg exchange reactions are not likely to quench at the same P-T conditions as geobarometers that are based on net-transfer reactions. Therefore, in these cases, P-T estimates deduced from thermobarometry using mineral compositions affected by late Fe-Mg exchange may give misleading results (e.g., Frost and Chacko, 1989; Spear and Florence, 1992; Spear, 1993). As found by Smith (1999), the accuracy of thermobarometric calculations for garnet peridotite xenoliths is best established for the two pyroxene thermometer plus Al-in-orthopyroxene barometer of Brey and Kohler (1990) and for P-T conditions in the range 20 to 50 kbar and 800 to 1100°C. Therefore, in contrast to previous workers (Stiefenhofer et al., 1999), we used the formulations of Brey and Kohler (1990) to make our basic thermobarometric estimates (Figure 10). However taking into account that the previously estimated ranges of pressure (up to ~70 kbar) and temperature (up to  $\sim 1400^{\circ}$ C) for mantle nodules from the Venetia kimberlitic pipes (Stiefenhofer et al., 1998) is partially outside of best established P-T region (Smith 1999), we also used alternative P-T estimates summarized in Table 3. For our thermobarometric calculations, we used the PTEXL program (a MS Excel® file program created by T. Koehler and A. Girnis). This program allows to proceeds with geothermobaromety of mantle rocks on the basis of compositions of coexisting minerals given in weight percent. Correction for Fe<sup>3+</sup> was taken into account as a part of computing procedure.

#### Results

The results of thermobarometry for four deformed and undeformed nodules are shown in Figure 10. As may be seen, most of nodules are characterized by significant variations in both pressure (42 to 74 kbar) and temperature (1260° to 1450°C) estimated using the composition of matrix minerals using the same combination of the two-pyroxene thermometer and garnet-orthopyroxene barometer of Brey and Kohler (1990). On the other hand, the maximum P-T estimates are consistent (70 to 74 kbar and 1400° to 1450°C) and may suggest similar initial conditions of crystallization of garnet-pyroxene assemblage. Variation in P-T estimates in most cases reflects simultaneous decrease in pressure and temperature from estimated peak conditions. This variation may, therefore, be attributed to reflect post-peak decompression/cooling history before the entrapment of the rocks by kimberlitic magma. This decompression and cooling proceeded either under low-strain (undeformed peridotites) or in highstrain (deformed peridotites) conditions. In this respect, formation of deformed peridotites might be genetically related to decompression, representing fragments of deep mantle shear zones controlling (convective?) exhumation of mantle rocks. The significant pressure drops recorded within all deformed nodules supports this possibility (Figures 10c and d). No significant temperature decrease occurred for an ~30 kbar drop in pressure in the deformed sample B-00-194 (Figure 11c). This may be attributed to a partial decoupling of the compositions of coexisting pyroxenes possibly due to kinetic effects (e.g., distinctly higher diffusion rates for the Fe and Mg compared to Ca and Al) hampering re-equilibration of these minerals during the fast exhumation process. On the other hand, a significant  $100^{\circ}$  to  $150^{\circ}$ C temperature decrease with decreasing pressure is recorded for this same sample using Fe-Mg garnetorthopyroxene (Harley, 1984), garnet-clinopyroxene (Ellis and Green, 1978; Powell, 1985; Krogh, 1988;) and garnet-olivine (O'Neil and Wood, 1978) Fe-Mg exchange thermometers (Table 3), suggesting a common decompression/cooling evolution for this sample (Figure 10c). Peak pressures of 68 to 74 kbar, estimated for Venetia peridotite samples, are outside of 45 to 60 kbar pressure range ("diamond window", Sobolev *et al.*, 2000) characteristic for garnet-orthopyroxene inclusions in diamond and peridotitic xenoliths containing diamond worldwide. However, taking into account relatively low ( $\pm$  10 to 15 kbar) accuracy of pressure calculations related to low Al<sub>2</sub>O<sub>3</sub> content in analysed orthopyroxenes (Table 2), this discrepancies might be related to the differences in the thermobarometers used.

Temperature estimates for the kelyphitic textures were also determined using the twopyroxene thermometer of Brey and Kohler (1990). These estimates are consistent for different samples and vary between 1140 to 1260°C at 15 kbar pressure. Lowest temperatures are recorded for the rims of coexisting minerals (Table 3) that corresponds to the cooling of nodules during the formation of kelyphite at the pressures <17 kbar (see intersection of isolines for two-pyroxene temperature estimates with garnet decomposition curve in Figure 10). Further cooling is recorded by the formation of serpentine whose stability field does not exceed 750°C in temperature (see Figure 10). For several samples, maximum temperatures recorded by kelyphite are consistent with minimum temperature recorded by matrix minerals (Figure 10b and d), which may suggest a continuous thermal evolution of peridotites before and after entrapment by kimberlitic magma. Together with the absence of annealing of deformation textures, the consistency of temperature estimates may suggest entrapment of nodules during or short after the shearing at a temperature ~1250°C. If this situation is not a coincidence, it may also suggest a link between deformation of mantle and the formation of the kimberlitic magma.

# Conclusion

The P-T evolution inferred from the garnet-harzburgite nodules studied is presented in Figure 11. Two interrelated stages of the evolution are suggested:

1) Pre-entrainment mylonitization followed by

2) Syn-entrainment decompression and subsequent cooling and hydration.

These two stages must have occurred rapidly and were very closely related in time because in the absence of stresses at the temperatures and pressures of mylonite and kelyphite formation, annealing to coarse-grained rocks would occur very rapidly if quenching did not take place (*e.g.* Passchier and Trouw, 1996; Smit and van Reenen, 1997).

It is also seen in Figure 11 that our P-T estimates deviate from the average mantle geotherm suggested by Stiefenhofer *et al.* (1999) for the nodules of Venetia kimberlitic pipes. Taking into account that we used thermometers and barometers that differ from those used by Stiefenhofer *et al.* (1998), the systematic deviation in P-T estimates must be mainly related to the discrepancies of different thermobarometric formulations (see Table 3). On the other hand relatively high temperatures estimated in studied nodules may result from the thermal disturbance of the steady state geotherm (e.g., Franz et al., 1996a, b) by the increased heat flow in the mantle during the kimberlite magmatism. This also coincide with intense contemporane-ous deformation of peridotites possibly related to the active mantle convection.

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Table 1. Mineral	textures and asser	nblages of garner	t-harzburgite nodules	s studied.

Sample	Rock type	Mineral assemblage
VN17	Undeformed, coarse-grained	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)
VN19	Undeformed, coarse-grained	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)
B-00-133	Moderately deformed, coarse-	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)
	grained	
B-00-134	Undeformed, coarse-grained	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)
B-00-136	Strongly deformed (mylonitized), medium-grained	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)
B-00-184	Moderately deformed, coarse- grained	Ol+Opx+Grt (with kelyphite)+Cpx+(Serp+Mag+Chl)

Table 2. Selected micro	nroha analyses of	f coovicting minor	baibute solution	
Table 2. Sciected intero	probe analyses of	coexisting inner	ais in samples studied.	

Sample			E	8-00-136									F	3-00-184								
Location		Mat	rix		k	Kelyphite	;				Ma	rix				Kelyphite						
Mineral	Ol	Opx	Срх	Grt	Opx	Срх	Spl	Ol	Ol	Opx	Opx	Срх	Срх	Grt	Grt	Opx	Срх	Срх	Spl	Spl		
								(core)	(rim)	(core)	(rim)	(core)	(rim)	(core)	(rim)		(core)	(rim)	(core)	(rim)		
Spot	B33	B37a	B35	B17c	B12a	B13a	B11a	N35	N28	N43	N29	N37	N42	N25	N26	N6	N1	N2	N3	N4		
										Weigh	nt %*											
SiO <sub>2</sub>	41.23	57.57	54.08	41.24	56.18	50.69	0.19	40.91	41.05	57.06	57.93	56.08	55.28	42.49	42.29	51.36	56.07	50.15	0.29	0.58		
TiO <sub>2</sub>	0.02	0.01	0.18	0.06	0.05	0.26	0.26	0.18	0.00	0.09	0.00	0.30	0.40	0.29	0.58	0.19	0.00	0.20	0.19	0.19		
$Al_2O_3$	0.05	0.87	1.66	16.45	2.49	6.60	37.43	< 0.11	0.00	1.30	1.19	2.21	2.21	21.49	20.03	10.11	2.21	9.02	46.58	46.83		
FeO	8.61	5.34	4.08	6.69	5.64	3.89	12.67	9.09	8.77	5.40	4.93	3.53	3.94	6.48	6.73	6.54	3.22	3.56	11.96	11.86		
MnO	0.12	0.16	0.18	0.29	0.25	0.23	0.27	0.18	0.09	0.19	0.47	0.40	0.20	0.67	0.29	0.47	0.30	0.20	0.29	0.19		
MgO	49.36	34.04	21.31	19.02	31.58	16.71	17.11	49.32	49.64	33.58	33.85	19.94	19.46	21.52	21.49	29.16	19.58	15.51	19.41	19.63		
CaO	0.11	1.39	17.01	7.31	2.49	18.63	0.12	0.00	0.07	1.28	1.12	15.59	15.95	4.73	4.44	1.66	17.31	20.18	0.08	0.23		
Na <sub>2</sub> O	0.03	0.05	0.38	0.00	0.07	0.50	0.02	0.06	0.06	0.12	0.18	0.65	0.65	0.06	0.06	0.12	0.45	0.39	0.00	0.06		
K <sub>2</sub> O	0.01	0.00	0.13	0.00	0.01	0.02	0.02	0.00	< 0.09	< 0.09	0.00	0.00	< 0.10	0.00	0.00	0.00	0.00	0.00	< 0.10	0.00		
$Cr_2O_3$	0.08	0.46	0.92	8.92	1.16	2.47	31.73	< 0.16	< 0.23	0.88	0.32	1.29	1.80	2.28	4.08	0.40	0.86	0.78	21.11	20.42		
NiO	0.38	0.11	0.07	0.02	0.07	0.00	0.19	n.a**.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.		
										Catior												
Si	1.006	1.983	1.949	3.012	1.948	1.841	0.005	0.999	1.001	1.969	1.990	1.998	1.981	3.012	3.013	1.786	2.000	1.815	0.008	0.016		
Ti	0.000	0.000	0.005	0.004	0.001	0.007	0.005	0.003	0.000	0.002	0.000	0.008	0.011	0.015	0.031	0.005	0.000	0.006	0.004	0.004		
Al	0.001	0.035	0.071	1.416	0.102	0.283	1.250		0.000	0.053	0.048	0.093	0.093	1.795	1.682	0.414	0.093	0.385	1.488	1.491		
Fe****	0.176	0.154	0.123	0.408	0.164	0.118	0.300	0.186	0.179	0.156	0.142	0.105	0.118	0.384	0.401	0.190	0.096	0.108	0.271	0.268		
Mn	0.002	0.005	0.006	0.018	0.007	0.007	0.006	0.004	0.002	0.006	0.014	0.012	0.006	0.040	0.017	0.014	0.009	0.006	0.007	0.004		
Mg	1.794	1.748	1.145	2.071	1.633	0.905	0.722	1.796	1.805	1.727	1.733	1.059	1.039	2.274	2.282	1.511	1.041	0.837	0.784	0.790		
Ca	0.003	0.051	0.657	0.572	0.093	0.725	0.004	0.000	0.002	0.047	0.041	0.595	0.612	0.359	0.339	0.062	0.662	0.783	0.002	0.007		
Na	0.001	0.004	0.027	0.000	0.005	0.035	0.001	0.003	0.003	0.008	0.012	0.045	0.045	0.008	0.009	0.008	0.031	0.028	0.000	0.003		
K	0.000	0.000	0.006	0.000	0.000	0.001	0.001	0.000	< 0.003	< 0.004	0.000	0.000	0.005	0.000	0.000	0.000	0.000	0.000	< 0.003	0.000		
Cr	0.002	0.013	0.026	0.515	0.032	0.071		< 0.003	< 0.004	0.024	0.009	0.036	0.051	0.128	0.230	0.011	0.024	0.022	0.452	0.436		
Ni	0.007	0.003	0.002	0.001	0.002	0.000	0.004	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.		
Sum	2.993	3.995	4.014	8.018	3.986	3.993	3.010	2.996	2.999	3.996	3.988	3.952	3.961	8.016	8.004	4.001	3.957	3.989	3.020	3.019		

Table 2. (continued)

$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Sample		·			VN17					VN19											
Spot         A51         A49         A50         A53         A9         A2         A5         A12         A11         H3         H36         H1         H15         H20         H16         H31         H31         H13         H13         H36         H1         H15         H20         H16         H31         H13         H14         H14         H15         H20         H16         H31         H13         H14         H15         H20         H16         H31         H13         H14         H14         H15         H20         H16         H31         H13         H14         H15         H20         H16         H31         H13         H14         H14         H15         H20         H16         H31         H13         H14         H15         H20         H16         H31         H13         H14         H15         H20         H16         H31         H34         H34         H33         H30         H33         H30         H33         H30         H31         H33         H30         H30         H30         H33         H34	Location		Ma	trix			ŀ	Kelyphite	e			Ma	trix				Kely	phite				
Spot         A51         A49         A50         A53         A9         A2         A5         A12         A11         H3         H36         H1         H15         H20         H16         H31         H13         H14           SiO2         41.48         57.36         55.45         42.50         53.94         49.85         48.98         0.09         0.09         41.37         57.92         54.18         41.48         49.94         49.91         46.54         46.32         0.10         0.16           Al2O3         0.00         0.98         1.99         20.15         6.04         8.17         8.61         57.18         52.63         0.00         0.76         1.52         18.97         10.02         9.76         10.77         11.96         46.89         48.22           FeO         8.73         5.20         4.05         7.20         6.30         7.94         50.37         1.98         21.04         28.96         28.56         14.58         13.81         19.25         19.20           CaO         0.07         1.43         17.03         4.44         2.13         18.65         17.71         0.00         0.08         0.15         1.05         18.64 <td< td=""><td>Mineral</td><td>Ol</td><td>Opx</td><td>Срх</td><td>Grt</td><td>Opx</td><td>Срх</td><td>Срх</td><td>Spl</td><td>Spl</td><td>Ol</td><td>Opx</td><td>Срх</td><td>Grt</td><td>Opx</td><td>Opx</td><td>Срх</td><td>Срх</td><td>Spl</td><td>Spl</td></td<>	Mineral	Ol	Opx	Срх	Grt	Opx	Срх	Срх	Spl	Spl	Ol	Opx	Срх	Grt	Opx	Opx	Срх	Срх	Spl	Spl		
							< /	· · /	` /	· /					```	< /	· /	< /	``	· · /		
$            SiO_2  41.48  57.36  55.45  42.50  53.94  49.85  48.98  0.09  0.09  41.37  57.92  54.18  41.48  49.94  49.91  46.54  46.32  0.10  0.10  0.10  0.00  0.00  0.38  0.38  0.71  1.01  0.37  0.38  0.00  0.19  0.51  0.58  0.38  0.38  0.38  1.22  1.32  0.10  0.10  0.10  0.10  0.00  0.09  0.19  0.00  0.97  1.52  18.97  10.02  9.76  10.77  11.96  46.89  48.22  1.32  0.10  0.19  0.10  0.19  0.09  0.19  0.00  0.19  0.01  0.10  0.17  1.10  0.37  0.38  0.00  0.76  1.52  18.97  10.02  9.76  10.77  11.96  46.89  48.22  0.10  0.19  0.09  0.19  0.00  0.19  0.38  0.30  0.40  0.37  0.19  0.09  0.38  0.10  0.29  0.47  0.29  0.61  0.51  0.19  0.38  0.00  0.09  0.19  0.00  0.19  0.38  0.30  0.40  0.37  0.19  0.09  0.38  0.10  0.29  0.47  0.29  0.61  0.51  0.19  0.39  0.38  0.10  0.29  0.47  0.29  0.61  0.51  0.19  0.39  0.38  0.00  0.20  0.07  1.43  13.44  1.48  1.92  1$	Spot	A51	A49	A50	A53	A9	A2	A5	A12				H36	H1	H15	H20	H16	H31	H13	H14		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$										V	Veight %	)										
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	-	41.48		55.45				48.98				57.92							0.10	0.10		
FeO         8.73         5.20         4.05         7.20         6.30         3.79         4.50         10.16         10.73         7.94         4.73         3.31         6.25         5.80         6.37         4.13         3.40         10.47         10.43           MmO         0.09         0.19         0.00         0.19         0.38         0.30         0.40         0.37         0.19         0.09         0.38         0.10         0.29         0.47         0.29         0.61         0.51         0.19         0.33           MgO         49.31         34.25         19.59         21.37         30.29         16.63         17.49         20.67         19.97         50.39         34.37         19.89         21.04         28.96         28.56         14.58         13.81         19.25         19.20           CaO         0.06         0.18         0.59         0.06         0.06         0.03         0.06         0.012         0.09         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00         0.00		0.09	0.00	0.00				1.01	0.37			0.19	0.51	0.58		0.38		1.32	0.10	0.19		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Al2O <sub>3</sub>	0.00	0.98	1.99	20.15	6.04	8.17	8.61	57.18	52.63	0.00	0.76	1.52	18.97	10.02	9.76	10.77	11.96	46.89	48.22		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	FeO	8.73	5.20	4.05	7.20	6.30	3.79	4.50	10.16	10.73	7.94	4.73	3.31	6.25	5.80	6.37	4.13	3.40	10.47	10.45		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	MnO	0.09	0.19	0.00	0.19	0.38	0.30	0.40	0.37	0.19	0.09	0.38	0.10	0.29	0.47	0.29	0.61	0.51	0.19	0.39		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	MgO	49.31	34.25	19.59	21.37	30.29	16.63	17.49	20.67	19.97	50.39	34.37	19.89	21.04	28.96	28.56	14.58	13.81	19.25	19.20		
K2O       <0.09       0.00       <0.10       <0.10       <0.10       <0.10       0.00	CaO	0.07	1.43	17.03	4.44	2.13	18.65	17.71	0.00	0.08	0.15	1.05	18.64	5.56	1.66	1.82	18.30	20.61	0.00	0.15		
Cr <sub>2</sub> O <sub>3</sub> 0.08       0.40       1.21       3.59       0.49       1.46       0.94       11.16       15.93       0.00       0.48       1.46       5.76       2.65       2.91       3.80       1.90       23.01       21.31         NiO       n.a.       n.a. <td>Na<sub>2</sub>O</td> <td>0.06</td> <td>0.18</td> <td>0.59</td> <td>0.06</td> <td>0.06</td> <td>0.33</td> <td>0.26</td> <td>0.00</td> <td>0.00</td> <td>0.06</td> <td>0.12</td> <td>0.39</td> <td>0.06</td> <td>0.12</td> <td>0.00</td> <td>0.07</td> <td>0.07</td> <td>0.00</td> <td>0.00</td>	Na <sub>2</sub> O	0.06	0.18	0.59	0.06	0.06	0.33	0.26	0.00	0.00	0.06	0.12	0.39	0.06	0.12	0.00	0.07	0.07	0.00	0.00		
NiO         n.a.	K <sub>2</sub> O	< 0.09	0.00	< 0.10	< 0.10	0.00	< 0.10	< 0.10	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	< 0.10	0.00	0.00		
Cations           Si         1.010         1.976         1.988         3.029         1.873         1.807         1.778         0.002         0.004         1.988         1.954         2.979         1.746         1.749         1.705         1.694         0.003         0.002           Ti         0.002         0.000         0.000         0.021         0.010         0.019         0.028         0.007         0.000         0.014         0.031         0.010         0.010         0.034         0.036         0.002         0.004           Al         0.000         0.040         0.084         1.692         0.247         0.349         0.368         1.743         1.636         0.000         0.014         0.031         0.010         0.013         0.465         0.515         1.492         1.527           Fe         0.178         0.150         0.121         0.429         0.183         0.115         0.137         0.220         0.237         0.161         0.136         0.100         0.375         0.169         0.187         0.126         0.104         0.236         0.234           Mn         0.002         0.006         0.000         0.012         0.008         0.002         <		0.08	0.40	1.21	3.59	0.49	1.46	0.94	11.16	15.93	0.00	0.48	1.46	5.76	2.65	2.91	3.80	1.90	23.01	21.31		
Si       1.010       1.976       1.988       3.029       1.873       1.807       1.778       0.002       0.004       1.988       1.954       2.979       1.746       1.749       1.705       1.694       0.003       0.003         Ti       0.002       0.000       0.000       0.021       0.010       0.019       0.028       0.007       0.000       0.005       0.014       0.031       0.010       0.010       0.034       0.036       0.002         Al       0.000       0.040       0.084       1.692       0.247       0.349       0.368       1.743       1.636       0.000       0.031       0.064       1.606       0.413       0.403       0.465       0.515       1.492       1.522         Fe       0.178       0.150       0.121       0.429       0.183       0.115       0.137       0.220       0.237       0.161       0.136       0.100       0.375       0.169       0.187       0.126       0.104       0.236       0.235         Mn       0.002       0.053       0.654       0.339       0.079       0.724       0.689       0.000       0.002       0.011       0.009       0.004       0.023       0.014       0.039       0	NiO	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.			n.a.	n.a.	n.a.								
Ti       0.002       0.000       0.000       0.021       0.010       0.019       0.028       0.007       0.000       0.005       0.014       0.031       0.010       0.010       0.034       0.036       0.002       0.004         Al       0.000       0.040       0.084       1.692       0.247       0.349       0.368       1.743       1.636       0.000       0.031       0.064       1.606       0.413       0.403       0.465       0.515       1.492       1.527         Fe       0.178       0.150       0.121       0.429       0.183       0.115       0.137       0.220       0.237       0.161       0.136       0.100       0.375       0.169       0.187       0.126       0.104       0.236       0.235         Mn       0.002       0.006       0.000       0.012       0.011       0.009       0.012       0.008       0.004       0.002       0.011       0.008       0.019       0.016       0.004       0.009         Mg       1.790       1.758       1.047       2.270       1.568       0.899       0.946       0.797       0.785       1.823       1.759       1.069       2.253       1.510       1.492       0.796       0											Cations											
Al0.0000.0400.0841.6920.2470.3490.3681.7431.6360.0000.0310.0641.6060.4130.4030.4650.5151.4921.527Fe0.1780.1500.1210.4290.1830.1150.1370.2200.2370.1610.1360.1000.3750.1690.1870.1260.1040.2360.235Mn0.0020.0060.0000.0120.0110.0090.0120.0080.0040.0020.0110.0030.0180.0140.0080.0190.0160.0040.009Mg1.7901.7581.0472.2701.5680.8990.9460.7970.7851.8231.7591.0692.2531.5101.4920.7960.7530.7750.766Ca0.0020.0530.6540.3390.0790.7240.6890.0000.0020.0040.0390.7200.4280.0620.0680.7180.8070.0000.004Na0.0030.0120.0410.0090.0040.0230.0180.0000.0000.0030.0080.0270.0990.0880.0000.006Ma0.0030.0120.0410.0090.0040.0230.0180.0000.0000.0000.0030.0080.0270.0090.0080.0000.0050.000Na0.0030.0040.0030.0000.0000.0	Si	1.010	1.976	1.988	3.029	1.873	1.807			0.002	1.004	1.988	1.954	2.979	1.746	1.749	1.705	1.694	0.003	0.003		
Fe0.1780.1500.1210.4290.1830.1150.1370.2200.2370.1610.1360.1000.3750.1690.1870.1260.1040.2360.235Mn0.0020.0060.0000.0120.0110.0090.0120.0080.0040.0020.0110.0030.0180.0140.0080.0190.0160.0040.009Mg1.7901.7581.0472.2701.5680.8990.9460.7970.7851.8231.7591.0692.2531.5101.4920.7960.7530.7750.765Ca0.0020.0530.6540.3390.0790.7240.6890.0000.0020.0040.0390.7200.4280.0620.0680.7180.8070.0000.000Na0.0030.0120.0410.0090.0040.0230.0180.0000.0030.0080.0270.0090.0080.0000.0050.0000.000K<0.003	Ti	0.002	0.000	0.000	0.021	0.010	0.019	0.028	0.007	0.007	0.000	0.005	0.014	0.031	0.010	0.010	0.034	0.036	0.002	0.004		
Mn         0.002         0.006         0.000         0.012         0.011         0.009         0.012         0.008         0.004         0.002         0.011         0.003         0.018         0.014         0.008         0.019         0.016         0.004         0.009           Mg         1.790         1.758         1.047         2.270         1.568         0.899         0.946         0.797         0.785         1.823         1.759         1.069         2.253         1.510         1.492         0.796         0.753         0.775         0.769           Ca         0.002         0.053         0.654         0.339         0.079         0.724         0.689         0.000         0.002         0.004         0.039         0.720         0.428         0.062         0.068         0.718         0.807         0.000         0.004           Na         0.003         0.012         0.041         0.009         0.004         0.023         0.010         0.000         0.000         0.003         0.008         0.027         0.008         0.000         0.005         0.000         0.000           K         <0.003		0.000	0.040	0.084	1.692	0.247	0.349	0.368	1.743	1.636	0.000	0.031	0.064	1.606	0.413	0.403	0.465	0.515	1.492	1.527		
Mg       1.790       1.758       1.047       2.270       1.568       0.899       0.946       0.797       0.785       1.823       1.759       1.069       2.253       1.510       1.492       0.796       0.753       0.775       0.765         Ca       0.002       0.053       0.654       0.339       0.079       0.724       0.689       0.000       0.002       0.004       0.039       0.720       0.428       0.062       0.068       0.718       0.807       0.000       0.004         Na       0.003       0.012       0.041       0.009       0.004       0.023       0.018       0.000       0.000       0.003       0.008       0.027       0.009       0.008       0.000       0.005       0.000       0.000         K       <0.003		0.178																		0.235		
Ca       0.002       0.053       0.654       0.339       0.079       0.724       0.689       0.000       0.002       0.004       0.039       0.720       0.428       0.062       0.068       0.718       0.807       0.000       0.004         Na       0.003       0.012       0.041       0.009       0.004       0.023       0.018       0.000       0.000       0.003       0.008       0.027       0.009       0.008       0.000       0.005       0.000       0.000         K       <0.003		0.002		0.000																		
Na         0.003         0.012         0.041         0.009         0.004         0.023         0.018         0.000         0.003         0.008         0.027         0.009         0.008         0.000         0.005         0.000         0.000         0.003         0.008         0.027         0.009         0.008         0.000         0.005         0.000         0.		1.790	1.758	1.047			0.899				1.823					1.492	0.796	0.753	0.775	0.769		
K       <0.003	Ca	0.002	0.053	0.654	0.339	0.079	0.724	0.689		0.002	0.004	0.039	0.720	0.428	0.062	0.068	0.718	0.807	0.000	0.004		
Cr         0.002         0.011         0.034         0.203         0.013         0.042         0.332         0.000         0.013         0.042         0.327         0.073         0.081         0.110         0.055         0.491         0.453           Ni         n.a.		0.003	0.012	0.041	0.009	0.004	0.023	0.018	0.000	0.000	0.003	0.008	0.027	0.009	0.008		0.005	0.005	0.000	0.000		
Ni n.a. n.a. n.a. n.a. n.a. n.a. n.a. n.a		< 0.003		< 0.005	< 0.009			< 0.005						0.000			0.000	< 0.005	0.000	0.000		
		0.002	0.011	0.034	0.203	0.013	0.042	0.027	0.228	0.332	0.000	0.013	0.042	0.327	0.073	0.081	0.110	0.055	0.491	0.453		
Sum 2.990 4.005 3.975 8.012 3.989 3.992 4.008 3.005 3.006 2.997 3.989 3.993 8.027 4.005 3.999 3.977 3.990 3.004 3.004	Ni	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.	n.a.		n.a.	n.a.	n.a.	n.a.		n.a.	n.a.	n.a.	n.a.		
	Sum	2.990	4.005	3.975	8.012	3.989	3.992	4.008	3.005	3.006	2.997	3.989	3.993	8.027	4.005	3.999	3.977	3.990	3.004	3.004		

\* all analyses are normalized to 100 wt.% \*\* n.a. – not analysed \*\*\* formulas are calculated: Ol and Spl – per 4 O; Opx and Cpx – per 6 O; Grt – per 12 O \*\*\*\* all Fe is calculated as Fe<sup>2+</sup>, correction for Fe3+ was taken into account for geothermobarometry as a part of standard pro-

cedure by using PTEXL program

Sample		N	linerals	3					J	Γ (°C) cal	culated	at P gi	ven				P (kbar)	calculate	ed at T	given	T and P calculated		
	Location	Ol	Opx	Срх	Grt	Р	Т	Т	Т	Т	Т	Т	Т	Т	Т	Т	Т	Р	Р	Р	T°C	P kbar	
						kbar	$[BK_T]$	$[KB_T]$	[Kr <sub>T</sub> ]	$[NW_T]$	[Ha <sub>T</sub> ]	$[EG_T]$	$[Po_T]$	$[We_T]$	$[BM_T]$	[Ta <sub>T</sub> ]		$[BK_P]$	$NG_P$ ]	$[MC_P]$	[BK <sub>T</sub> ]	$[BK_P]$	
VN17	Matrix	a59	a58	a57	a52	40			1140	1277		1197		1249	1325	1422	1100	47	48	49	1400	67	
						80	1429	1470	1333	1414	1455	1365	1353	1249	1386	1422	1500	74	71	75			
VN17	Matrix	A51	A49	A50	A53	40	1355	1234	1296	1175	1079	1320	1319	1268	1321	1435	1100	47	48	48	1428	70	
						80		1472		1322	1336	1502	1499	1268	1381	1435	1500	75	70	73			
VN17	Matrix	A54	A55	A50	A53	40	1355	1352	1296	1184	1079	1320	1319	1268	1321	1435	1100	47	48	48	1428	70	
						80			1509	1330		1502			1381	1435	1500	75	70	73			
VN17	Matrix	A54	A60	A57	A52	40	1338	1232	1129	1207	1121	1178	1167	1256	1329	1427	1100	48	49	46	1405	70	
						80	-	1470				1342			1391	1427	1500	76	71	71			
VN17	Matrix	A54	A56	A50	A53	40		1352		1184	1071	1320	1319	1264	1318	1434	1100	46	47	46	1423	68	
						80		1609				1502				1434	1500	73	69	71			
VN17	Matrix	A59	A60	A57	A22	40			1129			1178			1329	1427	1100	48	49	46	1405	70	
						80		1470	1317	1350	1385	1342	1329	1256	1391	1427	1500	76	71	71			
VN17	Kelyphite		A9	A2		10	1212							1185		1435							
						30	1260							1185		1435							
VN17	Kelyphite		A10	A2		10	1198							1216	1204	1483							
						30	1246							1216	1231	1483							
VN17	Kelyphite		A9	A5		10	1245							1219	1250								
						30	1296							1219		1452							
VN17	Kelyphite		A10	A5		10	1231							1251		1496							
						30	1282							1251	1270	1496							
VN19	Matrix	H3	H38	H36	H1	40		1334		1232	1117		1293		1262	1381	1100	55	56	51	1379	74	
						80		1587		1370					1319	1381	1500	82	78	77			
VN19	Matrix	H39	H38	H37	H2	40		1220				1332				1391	1100	54	56	51	1393	74	
						80		1456				1512			1341	1391	1500	81	78	77			
VN19	Matrix	H21	H42	H41	H26	40		1334				1252			1293	1402	1100	42	45	48	1346	56	
						80		1588				1423			1353	1402	1500	65	64	73			
VN19	Matrix	H25	H42	H40	H1	40	1273	1211	1125	936	1196	1157	1145	1200	1259	1375	1100	43	45	48	1303	55	
						80		1446				1316				1376	1500	66	64	73			
VN19	Matrix	H21	H22	H40	H1	40		1327		942		1157				1376	1100	41	44	46	1316	53	
						80			1305	1100		1316				1376	1500	64	62	71			
VN19	Matrix	H25	H24	H41	H2	40	1334	1218	1269	1004	1001	1278	1274	1237	1291	1400	1100	50	52	51	1397	68	
						80		1453	1470	1160	1241	1452	1446	1237	1350	1400	1500	75	73	77			
VN19	Kelyphite		H15	H16		10	1230							1193	1208	1494							
						30	1283							1193		1494							
VN19	Kelyphite		H12	H16		10	1225							1191	1208	1492							
							1278							1191	1234	1492							
VN19	Kelyphite		H20	H31			1113							1089	1080	1395							
						25	1160							1089	1103	1395							

Table 3. Results of thermobarometry of samples studied.

Table 3. (continued)

Sample		N	linerals	5					]	Г (°C) cal	culated	at P g	iven				P (kbar)	calcula	ted at T	given	T and P c	alculated
	Location	Ol	Opx	Cpx	Grt	Р	Т	Т	Т	Т	Т	Т	Т	Т	Т	Т	Т	Р	Р	Р	T°C	P kbar
						kbar	$[BK_T]$	[KB <sub>T</sub> ]	[Kr <sub>T</sub> ]	$[NW_T]$	[Ha <sub>T</sub> ]	$[EG_T]$	$[Po_T]$	$[We_T]$	$[BM_T]$	[Ta <sub>T</sub> ]	°C	$[BK_P]$	$[NG_P]$	$[MC_P]$	$[BK_T]$	$[BK_P]$
B-00-136	Matrix	B33	B32	B31	B18a			1280					1321		1175	1324	1100		49	51	1263	53
								1525		1346	1377	1491	1489	1119	1225	1324	1500	65	65	78		
B-00-136	Matrix	B33	B37	B36	B17c			1285			1099	1262			1249	1379	1100	47	51	51	1324	58
								1531			1355				1305	1379	1500	68	67	77		
B-00-136	Matrix	B33	B26	B25	B17b			1303			1121				1340	1443	1100		56	49	1421	70
								1552				1449	1444		1402	1443	1500		74	75		
B-00-136	Matrix	B33	B37a	B35	B17c			1305			1106	1337	1338	1279	1332	1434	1100		47	49	1397	58
							1450		1585	1333	1363	1509	1509	1279	1393	1434	1500		63	75		
B-00-136	Matrix	B33	B32	B31s	B18a			1280	1405	1211	1119	1352			1213	1366	1100	45	49	51	1303	55
								1526	1607	1346	1377	1526	1527	1157	1266		1500	65	65	78		
B-00-136	Kelyphite		B20	B23			1222							1142	1193	1418						
							1274							1142	1218							
B-00-136	Kelyphite		B12a	B13a			1220							1160	1201	1404						
							1269							1160	1227							
B-00-136	Kelyphite		B14	B10a			1188							1126		1375						
							1238							1126		1375						
B-00-184	Matrix	N35	N43	N37	N25		1379	204	1279		1230	1300		1318	1378	1462	1100	47	47	44	1451	70
							1475	313	1487		1515		1475		1442	1462	1500		69	67		
B-00-184	Matrix	N28	N34	N38	N26	40	1386	1251	1144		1148	1195			1391	1468	1100		41	45	1429	59
							1479	1492	1335		1418	1363	1351		1457		1500		61	69		
B-00-184	Matrix	N36	N33	N39	N12		1387	205	1275		1176				1384	1472	1100		30	45	1395	43
							1483	313	1481		1450		1463		1449		1500		47	69		
B-00-184	Matrix	N28	N29	N42	N26		1381	1243	1341				1357		1352		1100		30	45	1385	42
							1480	1482	1560	1412	1371	1541	1542		1415		1500	47	47	69		
B-00-184	Kelyphite		N21	N24			1216							1276		1548						
							1265							1276		1548						
B-00-184	Kelyphite		N22	N24			1227							1240		1509						
							1276							1240		1509						
B-00-184	Kelyphite		N6	N2			1119							1127	1097							
							1165							1127	1120							
B-00-184	Kelyphite		N7	N1			1235							1377	1296							
						30	1280							1377	1327	1539						

 $\begin{array}{l} Geothermometers used: [BK_T] - Opx-Cpx \ (Brey and K\"ohler, 1990); [Kr_T] - Cpx-Grt \ (Krogh, 1988); [NW_T] - Ol-Grt \ (O'Neil and Wood, 1979); [KB_T] - Ol-Cpx \ (K\"ohler and Brey, 1990); [Ha_T] - Opx-Grt \ (Harley, 1984); [EG_T] - Cpx-Grt \ (Ellis and Green, 1979); [Po_T] - Cpx-Grt \ (Powell, 1985); [We_T] - Opx-Cpx \ (Wells, 1977); [BM_T] - Opx-Cpx \ (Bertrand and Mercier, 1985); [Ta_T] - Opx-Cpx \ (Taylor, 1998). \end{array}$ 

Geobarometers used: [BK<sub>P</sub>] – Opx-Grt (Brey and Köhler, 1990); [NG<sub>P</sub>] – Opx-Grt (Nickel and Green, 1985); [Mc<sub>p</sub>] – Opx-Grt (McGregor, 1974).



Figure 1: Map showing the location of the Venetia kimberlite cluster within the tectonic units of southern Africa (modified from Barton and Pretorius, 1998).



Figure 2: Map showing the location of eleven of the thirteen kimberlite bodies comprising the Venetia cluster within the regional geology. The bodies are oriented using the local north-south grid and each square is  $200 \text{ m}^2$ .



Figure 3: Map showing the kimberlite types presently exposed within the K1 pipe. The bodies are oriented using the local north-south grid and each square is 100 m2. TKB = tuffacitic kimberlite breccia. H = hypabyssal kimberlite. The majority of the nodules in the RAU collection were collected from H-N (hypabyssal north) and H-S (hypabyssal south) on levels 7 through 8 with some from levels 9 and 10.



Figure 4: Nodules studied for this manuscript. (A, B and C) VN17, VN19 and B-00-134 respectively; unsheared. Note kelyphitic rims around garnet grains and randomly oriented crystals of orthopyroxene and olivine. In VN-19, note the dark rim around nodule where orthopyroxene and olivine have been serpentinized by reaction with kimberlite magma. In (A), the lack of this rim shows that it is a fragment of a larger nodule. (D and E) B-00-184 and B-00-133 respectively; moderately deformed. Note the winged garnet in the upper left hand corner of E and the kelyphitic rim around it denoting that the rim formed after deformation. Note also the preferred orientation of the orthopyroxene crystals. (F) B-00-136 strongly deformed. Note the linear fabric defined by crystals of orthopyroxene and garnet and the rim around the nodule resulting from reaction with the kimberlitic magma. Olivine grain size has been strongly reduced so that individual grains are not obvious. Note also the smaller size of the strongly sheared nodule compared



Figure 5: Back-scattered electron images of kelyphitic textures in the samples studied.



Figure 6: Back-scattered electron images of deformation textures in sample B-00-136.

(*a*) Sigmoid shaped porphyroblast of garnet. (*b*) Delta/sigmoid shaped porphyroclast of orthopyroxene in fine-grained matrix of mylonite composed of olivine and serpentine. (*c*) Ribbon orthopyroxene and garnet porphyroblast in fine-grained matrix of mylonite composed of olivine and serpentine.(*d*) Deformed porphyroblast of orthopyroxene surrounded by olivine and serpentine.



Figure 7: Examples of hydration textures developed in the samples studied.



Figure 8: Chemical composition (a) and (b) and zoning (c) and (d) of garnets in the samples studied.



Figure 9: Chemical compositions of matrix (a) and (b) and kelyphite (c) and (d) pyroxenes in the samples studied.



Figure 10: Results of geothermobarometry of samples studied. Thin lines are calculated at the basis of compositions of coexisting minerals using two-pyroxene thermometer (steep lines) and garnet-orthopyroxene barometer (gently sloping lines) of Brey and Kohler (1990). Symbols show individual P-T estimates. Arrows show possible crystallization/deformation (dashed arrows) and exhumation (dotted arrows) trajectories for individual samples. Garnet and serpen-

32



Figure 11: Generalized scheme of the P-T history inferred for studied peridotites. Symbols show individual P-T estimates for different samples calculated using two-pyroxene thermometer and garnet-orthopyroxene barometer of Brey and Kohler (1990). Arrows show possible crystallization/deformation (dashed arrow) and exhumation (dotted arrow) generalized trajectories. Garnet and serpentine stability fields are after Schmidt and Poli (1999). Geotherm suggested by Stiefenhofer *et al.* (1999) is shown for comparison.